Report to the Grant AOARD-08-4049

Development of High ZT Thermoelectric Materials With Nanostructure For Energy Applications

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I. Background and motivation:

Thermoelectricity was first observed in 1823 by Seebeck, who found that a current would flow in a closed circuit made of two dissimilar metals when the two junctions were maintained at different temperatures. Twelve years later, a reverse effect was discovered by Peltier, who observed temperature changes in the vicinity of the junction between dissimilar conductors when a current was passed through the junction. In 1851 W. Thomson established a relationship between the Seebeck coefficient (S) and Peltier coefficient (Π) and predicted the existence of a third thermoelectric effect, the Thomson effect, which he subsequently observed experimentally. Since then much work has been done to make use of thermoelectric physics. Thermoelectric materials have the advantage of no mechanical parts, no need for coolant to operate, quiet performance, and can be utilized as a heat-electricity converter. Therefore, thermoelectric materials can be used either as a power generator based on the Seebeck effect, or as a thermo-cooler based on the Peltier effect. The current state-of-the-art thermoelectric materials commercially used for refrigeration and power generation are Bi_2Te_3 with T down to about 250 K and $Si_{1-x}Ge_x$ with T > 700 K.

II. The efficiency of thermoelectric materials- figure of merit ZT:

The efficiency of thermoelectric materials is determined by the dimensionless figure of merit ZT $ZT = S^2 \sigma T / (\kappa_e + \kappa_1)$,

where S is the Seebeck coefficient ($S = \Delta V/\Delta T$), σ is the electrical conductivity, κ_e is the thermal conductivity of electrons, and κ_l is the thermal conductivity of lattice phonons. Good thermoelectric materials should possess a large Seebeck coefficient, low thermal conductivity $\kappa_e + \kappa_l$, and high electrical conductivity. This concept is related to Slack's earlier proposal of a "Phonon-Glass/Electron-Crystal" (PGEC) model, which suggests that a good thermoelectric material should have the electronic properties of a crystalline material and the thermal properties of a glass. Applications only become practical for ZT > 2, and at a ZT of 3, thermoelectric materials would become competitive with vapor-compression refrigeration systems. It has been a challenge to increase ZT > 1 since the parameters, s, σ and κ , are generally interdependent.

By properly adjusting the parameters of S, σ and κ , the figure of merit ZT can possibly be enhanced. According to *Weideman-Franz law*, $\sigma T/\kappa_e = 3(k/e)2/\pi^2 = 1/(156 \ \mu v/K)^{-2}$ is a constant. This means that in general, increasing the electrical conductivity σ will increase κ_e proportionally. We ignore the lattice thermal conductivity, κ_l , as it can be reduced by increasing phonon scattering.

Proposed Works

There are three topics proposed as the core researches in this proposal. In the following we describe, respectively, our research results on the proposed topics.

Research approach A—Investigate the systems of "surface modified nanostructured bulk"

Report Documentation Page

Form Approved OMB No. 0704-0188

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1. REPORT DATE 03 MAR 2010	2. REPORT TYPE Final	3. DATES COVERED 25-03-2008 to 05-03-2010	
4. TITLE AND SUBTITLE Development of High ZT Thermoelectric Materials with Nanostructure for Energy Applications		5a. CONTRACT NUMBER FA48690814049	
		5b. GRANT NUMBER	
	5c. PROGRAM ELEMENT NUMBER		
6. AUTHOR(S) Maw-Kuen Wu		5d. PROJECT NUMBER	
		5e. TASK NUMBER	
	5f. WORK UNIT NUMBER		
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Academia Sinica,128 Academia Rd., Section 2,Nankang,Taipei ,TW,11529		8. PERFORMING ORGANIZATION REPORT NUMBER N/A	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES) Asian Office of Aerospace Research & Development, \((AOARD)\), Unit 45002, APO,		10. SPONSOR/MONITOR'S ACRONYM(S) AOARD	
AP, 96338-5002		11. SPONSOR/MONITOR'S REPORT NUMBER(S) AOARD-084049	

12. DISTRIBUTION/AVAILABILITY STATEMENT

Approved for public release; distribution unlimited

13. SUPPLEMENTARY NOTES

14. ABSTRACT

Three subtasks were performed in this research study and the respective findings are described here. A?Investigation of the system of surface modified nanostructured bulk: Our studies show that the CuFeSe2 thin film on Si (or SiO2) substrates exhibit large thermal power in the order of 1300 microvolt/K at ~300K. We conclude that the thinner films have the higher thermal conductivity. At 300K, the magnitude of thermal conductivity 8.8 W/m-K for 200 nm is about two times larger than 1.7 W/m-K of the 800 nm film. The consequence is attributed to the better crystallization in thinner films. This film thickness dependence of crystallization was confirmed by X-ray diffraction data. In this report, the thermal resistance of the interface boundaries in SiNx-CuFeSe2 was estimated to be 3.32?0-8 Km2/W. We cannot give a clear estimate of the ZT value for these films at the moment. The major difficulty came from the exact value of the thermoelectric power of the thin film. B?Investigation of one-dimensional nanowires: Bismuth telluride film and nanowires array (embedded in alumina template) were fabricated by potentiostatically electrodeposition. The Seebeck coefficient, electrical resistivity and thermal conductivity of a single nanowire provide an estimated thermoelectric figure of merit ZT = 0.45 at 300 K and likely to reach beyond 0.9 above 350 K, for Bi2Te3 nanowires. C?Directional dependent thermal conductivity and thermal rectifier: we cannot make a conclusive statement that there exists no rectification effect, as the narrowest section in our device is still too big compared with that observed in carbon and boron nitride nanotubes.

15. SUBJECT TERMS

Nanotechnology, Thermoelectric, Figure of merit ZT

16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF	18. NUMBER OF	19a. NAME OF
			ABSTRACT	PAGES	RESPONSIBLE PERSON
a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified	Same as Report (SAR)	18	RESI ONSIBEL I ERSON

In this particular topics, we have put our focus on the investigation of a newly synthesized materials, CuFeSe₂, which has a layer-like structure. In addition, it also exhibit an interesting magnetic characteristic originates from the Fe-Se layer. Therefore, we have carried out an extensive study on the the preparation and characterization of CuFeSe₂ compound, in particular the material in thin film form.

CuFeSe₂ is a type I-III-VI₂ semiconductor with direct band gap 0.16 eV [1]. It has a tetragonal structure, with space group P-42c and lattice constants a = 5.53 Å, c = 11.049Å [2]. It is considered to be the small band gap photoelectric material [3-6]. For semiconductor applications the thermal properties of thin film are important parameters. We used the pulse laser ablation (PLA) to grow high quality CuFeSe₂ thin film on silicon oxide substrate, and use X-ray diffraction to determine the its structure and quality. The result shows that the thin films have highly preferred orientation. Here we applied the differential 3ω technique to the measurement of thermal conductivity for different thickness thin film specimens, to extract the thermal boundary contribution and the intrinsic thermal conductivity of the thin films.

A1. Sample preparation

Thin films were grown by laser ablation from a target with the composition Cu:Fe:Se = 1:1:2, the target is 25.4 mm in diameter and 8 mm in thickness. The target was sintered at $600 \sim 700$ °C under vacuum. After sintering, the target was mechanical polished before each deposition process in order to keep the consistence and the reproducibility for successive deposition processes. Thin films were deposited in vacuum of 10^{-5} Torr. A KrF excimer laser with $\lambda = 248$ nm (Lambda Physik LPX Pro) was chosen as the ablation source. The laser beam was scanned on the surface of target with pulse frequency of 5 Hz. The scanning area is a $10 \times 4 \text{ mm}^2$ rectangle. The laser power is $5 \sim 6 \text{ J/cm}^{-2}$. The distance between the target and the substrate is around 4.5 to 5.0 cm. In this work, thin films were deposited on $12 \times 12 \text{ mm}^2$ Si wafer (1 0 0) substrates with 400 nm SiO₂ surface layer at 250 °C. Before the deposition, the substrates were ultrasonically cleaned in methanol, isopropanol, acetone, and then fixed by silver paint on the stainless steel holder. For differential 3ω measurement purpose, the thin films was deposited on one half of the substrate with the rest half for reference.

On top of the $CuFeSe_2$ film a 300 nm SiN_x thin layer was then deposited using plasma-enhanced chemical vapor deposition (PECVD) to serve as an insulation layer for the later deposited Au-Cr strip. We chose SiN_x for the insulation layer because of its good thermal conductivity. The Au-Cr strip was first thermal deposited a 10 nm Cr film then followed by a 150 nm Au film, the width and the length of the Au-Cr microprobe is 4 μ m and 1.5 mm respectively. There are two microprobes, one on the thin film and the other on the substrate (Fig. A1). With these two microprobes the temperature drop across the thin film can be measured, and the thermal conductivity can be derived afterwards.

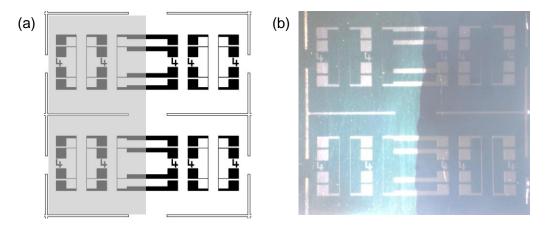


Fig. A1 (a) The schematic diagram shows the microprobe pattern, the gray side represents the film deposited (b) The corresponding specimen, in which CuFeSe₂ film was deposited in the left half (the bright part), the right half is the reference without film deposited.

A2. Structural characterization

The CuFeSe₂ films were characterized by X-rays diffraction using a PANalytical X'Pert PRO MPD system with copper anode ceramic tube as source and multi channels X'Celerator as detector. The theta-2theta measurement results are shown in Fig. A2. The preferred orientation in (h 0 0) planes was clearly reflected in (200), (400), and (600) peaks. This means the crystalline in the films are highly aligned, and with a-b plane perpendicular to the substrate surface. Particularly a small peak of CuFeSe₂ (112) starts to grow for 400 nm film.

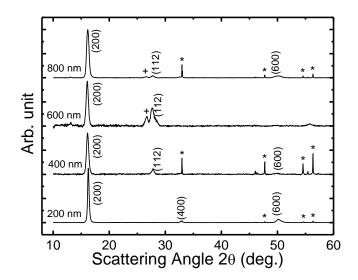


Fig. A2 The x-ray diffraction for 200, 400, 600 and 800 nm films.

The planes of (200), (400), and (600) are clearly observed in 200 and 400 nm films. The mark \ast represents the peak of SiO₂ substrate. The mark + represent the second phase Cu₂Se (111) peak in 600 and 800 nm films. It shows the 800 nm and 600 nm thin films have larger d-spacing and orientation distribution.

The similar result was also shown in the rocking curve measurement data (Fig. A3). The rocking curve of (200) plane ($2\theta=16.104^{\circ}$) were measured for 200 nm, 600 nm, and 800 nm films. It is very clear that 800 nm and 600 nm sample have the broader width than 200 nm sample. Although the thicker CuFeSe₂ film has preferred orientation in (200) but the orientation of the crystalline is not as fine as it in 200 nm sample. This explains why the lower cross plane thermal conductivity in thicker films.

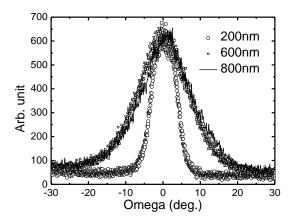


Fig. A3 The rocking curve profile for 200, 600, and 800 nm sample. The 200 nm sample has the thinner width of half maximum than thicker samples.

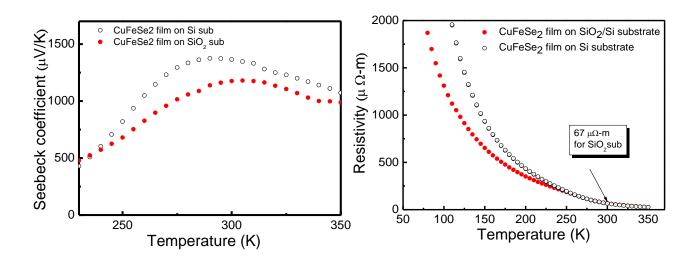


Figure A4. Temperaure dependence of Seebeck coefficient (left panel) and electrical resistivity (right panel) of CuFeSe₂ thin films (on bot Si and SiO₂ substrates).

A3. 3\omega technique for thin film

The thermoelectric power and electrical resistivity of the deposited film are shown in Figure A4. The thermoelectric power of this material shows a broad maximum at around 300K. The value is unexpectedly high on either Si or SiO₂ substrartes. On the other hand, a strong semiconductor-like resistivity was observed in these materials. The most difficult part in this study is to determine the thermal conductivity of the films. The followings are the results of our study.

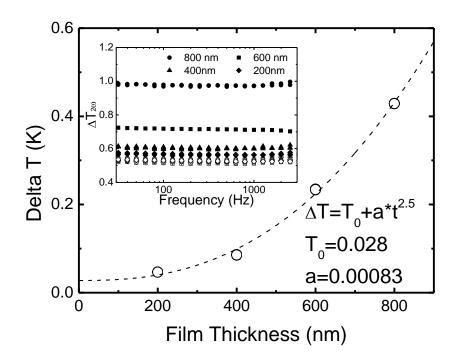


Fig. A5 The thickness dependence of temperature drop across film. Inset: The solid and open symbols represent the temperatures of top and bottom surfaces of films respectively at 300 K.

The thermal conductivity across films was measured by differential 3ω technique [7,8]. A strip consisting layers of 150 nm Au and 10 nm Cr with 4 µm width and 1.5 mm length is deposited as the microprobe by photolithography. The microprobe was driven by an AC power P at frequency ω . The resulting third-harmonic voltage was measured by a lock-in amplifier. The 3ω signals both on thin film

and reference part are converted to AC temperature variation. The difference between two microprobes is the temperature drop across CuFeSe₂ thin film, which is used to calculate the cross plane thermal conductivity.

The temperature drop of different thickness film was shown in Fig. A5. By the thickness dependence of temperature drop, it is clearly seen that the thicker film samples of 600and 800 nm, have higher temperature drop which is attributed to the structural disorders in films. This result is consistent with the conclusion of X-ray data, i.e. the thinner films have better crystallization in structure. A more detailed analysis of the temperature drop shows it fits to the equation $\Delta T = \Delta T_0 + a t^n$, where t represents the film thickness. The thermal resistance K_B of the interface boundaries was estimated as $3.32 \times 10^{-8} \text{ K-m}^2/\text{W}$ from $P/\Delta T_0$. Applying this result to all films, the thermal conductivity of all the films was obtained and plotted in Fig. A6 and A7. Basically the thermal conductivity is approximately linear on temperature. One interesting finding is that the thinner of the film the larger the thermal conductivity. For example, at 300 K the thermal conductivity of 200, 400, 600 and 800 nm films are 8.8, 5.7, 2.4, and 1.7 W/m-K respectively. This consequence is attributed to the condition of crystallization which was confirmed by X-ray diffraction data Fig. A3.

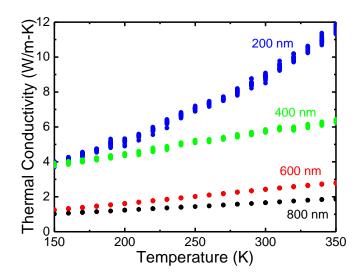


Fig. A6 Temperature dependence of thermal conductivity of 200, 400, 600, and 800 nm $CuFeSe_2$ thin films.

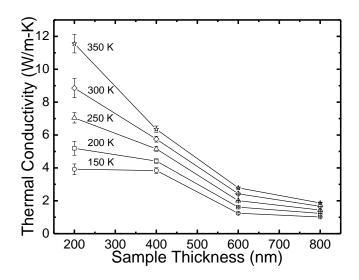


Fig. A7 The thickness dependence of thermal conductivity at various temperatures.

The inset in Fig A5 shows the original data of frequency dependence of temperature variation $\Delta T_{2\omega}$ of sample and reference respectively. Their difference is the temperature drop delta T across the film. The thermal conductivity of Si substrate was calculated from the slope of $\Delta T_{2\omega}$ vs. frequence. [7]. The result is consistent with that reported in literature.

A4. Summary of our results

Our studies show that the CuFeSe $_2$ thin film on Si (or SiO $_2$) substrates exhibit large thermal power in the order of 1300 μ V/K at ~300K. However, it decreases rapidly as temperature drops. Unfortunately, we do not have the setup for measurement at temperature above 350K. It will be interesing to find out how it behaves at this elevated temperature. On the other hand, the electrical resistivity shows typical semiconductor-like behavior with a value close to 67 μ D-m at 300K. We also obtained the thermal conductivity of the CuFeSe $_2$ thin film on silicon oxide substrate. We conclude that the thinner films have the higher thermal conductivity. At 300K, the magnitude of thermal conductivity 8.8 W/m-K for 200 nm is about two times larger than 1.7 W/m-K of the 800 nm film. The consequence is attributed to the better crystallization in thinner films. This film thickness dependence of crystallization was confirmed by X-ray diffraction data. In this report, the thermal resistance of the interface boundaries in SiN $_x$ -CuFeSe $_2$ was estimated to be 3.32×10^{-8} Km 2 /W. We can not give a clear estimate of the ZT value for these films at the moment. The major difficulty came from the exact value of the thermoelectric power of the thin film. Our measurements give only the results of the films plus the contributions from the substrate. We have the difficulty to separate these two contributions with our current experimental set-up.

Research approach B—Investigate one-dimensional nanowires

The Si nanowires of 9-20 nm and 50-200 nm investigated by Boukai et al and Hochbaum et al, respectively, showed smaller thermal conductivity, larger Seebeck coefficient and an enhancement of ZT~1 at room temperature. The 100-fold improvement over bulk Si is really a breakthrough in thermoelectric nanowires. Intrinsic Bi₂Te₃ bulk is n –type with S~250 $\mu v/K$, whereas bulk Sb₂Te₃ is p-type with S = 100 $\mu v/K$. Our preliminary studies on Bi₂Te₃ and Sb₂Te₃ nanowire arrays in anodized alumina template revealed that the Seebeck coefficients are 60 $\mu v/K$ for 50 nm Bi₂Te₃ and 160 $\mu v/K$ for 70 nm Sb₂Te₃ nanowire arrays. The Seebeck coefficient of the former is smaller than its bulk, whereas the latter shows opposite result with Seebeck coefficient larger than the bulk. The size effects on the two systems are quite different and should be further investigated.

To elucidate low-dimensional effects on thermoelectric materials, with the support of this grant, we further investigated the bismuth telluride film and nanowires array, which were fabricated by potentiostatically electrodeposition. The followings are our reports on the synthesis and characterizations of these nano-materials.

B1. Synthesis of the nanomaterials

 Bi_2Te_3 films were electrodeposited potentiostatically at -150 mV onto an ITO glass substrate from 1 M HNO₃ solution containing 0.008 M Bi^{3+} and 0.014 M HTeO₂⁺. The overall chemical reaction can be described as [9]

$$2 \text{ Bi}^{3+} + 3 \text{ HTeO}_{2}^{+} + 9 \text{ H}^{+} + 18 \text{ e}^{-} \rightarrow \text{Bi}_{2}\text{Te}_{3(s)} + 6 \text{ H}_{2}\text{O}$$
. (2)

The process was carried out in a conventional three-electrode cell with a saturated calomel electrode as the reference and a platinum sheet as the counter electrode. To remove oxygen from solution, argon was bubbled into the electrolyte throughout the experiment. After electrodeposition, asprepared Bi_2Te_3 films were annealed in argon with 5 % H_2 at 523 K for 2 hours.

The AAM template was produced in a two-step anodized process [10]. An aluminum foil with purity 99.999 % and thickness 0.13 mm (Alfa Aesar) was anodized in acid solutions, yielding an array of densely packed, hexagonally arranged, parallel nanochannels. The channel diameter can be

systematically controlled by anodized voltage, type of electrolyte and bath temperature. After anodization, the remaining aluminum layer was removed by CuCl₂ solution. A subsequent etching treatment by 0.5M NaOH aqueous solution for 7~10 min was carried out to remove the barrier layer on the bottom side, followed by depositing a gold-film layer to serve as conductive electrode.

By filling the template, a Bi₂Te₃ nanowires array was then fabricated with similar growth conditions, rinsed with deionized water and dried in nitrogen. A thin gold film (~300 nm) was thermally evaporated onto the top surface for oxidation prevention. It was suggested that annealing treatment to nanowires would improve their thermoelectric properties [11]. Our experience indicated that annealing does lower electrical resistivity of as-prepared film as expected, but the annealing-induced thermal shock leads to extremely high electrical resistance in nanowires embedded in AAM. Therefore, experimental data and analysis were confined to Bi₂Te₃ bulk, annealed films and as-prepared nanowires. To obtain individual nanowires, AAM was dissolved in 2 M NaOH aqueous solution, then washed by deionized water and centrifuged to remove residual NaOH thoroughly. Separated Bi₂Te₃ nanowires were dispersed in ethanol, ready for measurements.

For comparison with films and nanowires, a Bi_2Te_3 bulk was prepared by Bridgeman method. Briefly, high-purity (99.999 %) bismuth and tellurium powers were mixed in the appropriate molar ratio and sealed in a quartz tube, which was evacuated to 10^{-5} torr, kept in a furnace at 1073 K for 12 hours, and then cooled down to room temperature in one day.

B2. Structure characterizations

A 3-KW diffractometer (Philips) equipped with an array detector based on real time multiple strip was employed for crystal structural characterization. Morphologies of films and nanowires array were determined by a field emission scanning electron microscope (SEM, Hitachi S-4200), and compositions by energy dispersive X-ray. Transmission electron micrographs (TEM) and diffraction patterns were obtained with a field emission transmission electron microscope operated at 200 kV (JEOL JEM-2100).

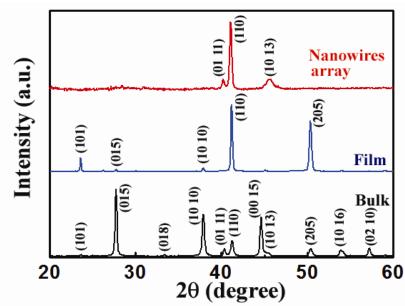


Figure B1. X-ray diffraction patterns for Bi₂Te₃ bulk, annealed film and as-prepared nanowires array.

X-ray diffraction patterns for Bi_2Te_3 bulk, annealed film and as-prepared nanowires array are presented in Figure B1. All peaks can be indexed to those of rhombohedral structure (R $\bar{3}$ m), confirming their single-phase purity [12]. The strong (110) peak suggests that the c-axis is parallel to the film surface and perpendicular to the axis of nanowire, respectively. Figure B2 shows the morphology of as-prepared and annealed Bi_2Te_3 films. Obviously, annealing yields formation of nanocrystalline structures. The scanning electron micrographs suggest a nucleation-and-growth

mechanism on surface of the substrate. Figure B3a reveals a hexagonally ordered nanochannels in AAM. The approximately 120-nm pore diameter would be then that of Bi₂Te₃ nanowires. Figure B3b is a mechanically polished top view of AAM template after being deposited with Bi₂Te₃. Only a few pores remain blank. A same conclusion can be made from the side view in Figure B3c, where most of the pores are filled with Bi₂Te₃. The filling ratio and porosity of the template are estimated to be about 88 % and 25 %, respectively, comparable to the best samples of this kind ever reported [13]. Separated nanowires in Figure B3d were obtained after AAM was dissolved by aqueous NaOH. Most of them have lengths longer than 20 µm. Judging from energy dispersive X-ray analysis, only elements present in both films and nanowires are Bi and Te in the ratio of 0.37: 0.63, corresponding to Te-rich n-type Bi₂Te₃. Extra Te may be situated as substitution to Bi sites or at interstitial sites. They may also be in grain boundaries, but not enough to be observed in XRD as impurity phase [14].

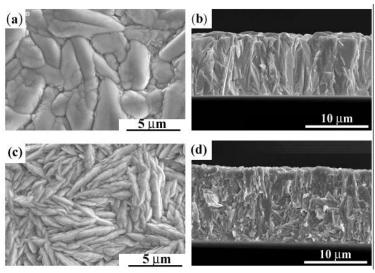


Figure B2. Scanning electron micrographs of Bi_2Te_3 films: (a) planar and (b) cross-sectional view of as-prepared film, (c) planar and (d) cross-sectional view of annealed film.

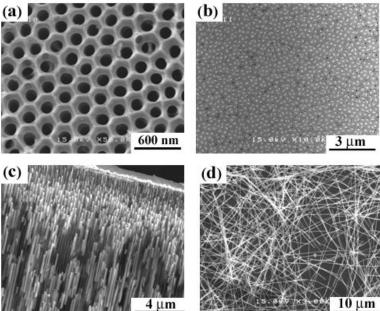


Figure B3. Scanning electron micrographs of AAM and Bi₂Te₃ nanowires array. (a) Highly ordered nanopores in AAM. (b) Top view of Bi₂Te₃ nanowires array. (c) Side view of Bi₂Te₃ nanowires array partially embedded in AAM. (d) Separated nanowires after AAM being dissolved.

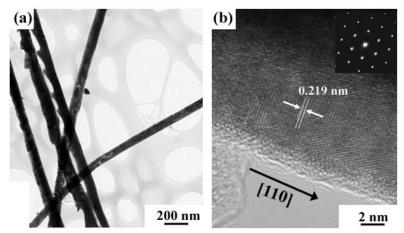


Figure B4. (a) TEM image of Bi₂Te₃ nanowires. (b) HRTEM image of a single Bi₂Te₃ nanowire. Inset: Diffraction patterns of a selected area.

TEM image in Figure B4a reveals the uniform profile of nanowires. A typical high resolution TEM (HRTEM) image of a single Bi₂Te₃ nanowire, along with electron diffraction patterns from a selected area, is given in Figure B4b. Lattice fringes yield an inter-planar distance of about 0.219 nm, consistent with the inter-planar spacing of the {110} planes of rhombohedral Bi₂Te₃ bulk. This result confirms the direction of the long axis of nanowire is [110].

B3. Properties characterizations

A standard four-probe resistivity measurement was performed on Bi₂Te₃ bulk, films and nanowires with a physical property measurement system (Quantum Design). However for nanowires array, gold layers were first deposited on the two sides of array employed as electric contacts of the four-probe method. Measurements of Seebeck coefficient between 120 and 350 K were performed by a longitudinal DC steady-state method on specially designed setups for film (Figure B5a) and nanowires (Figure B5b), with $S = \Delta V/\Delta T$, where ΔT is the temperature gradient across sample, and ΔV the thermo-emf generated by ΔT . Typically, ΔT is set in the range of 0.5-1 K monitored by a pair of T-type thermocouples. The resulting thermo-emf ΔV was continuously recorded by a digital voltmeter. By varying ΔT , a statistically averaged S value was derived from the slope of a ΔV vs. ΔT plot.

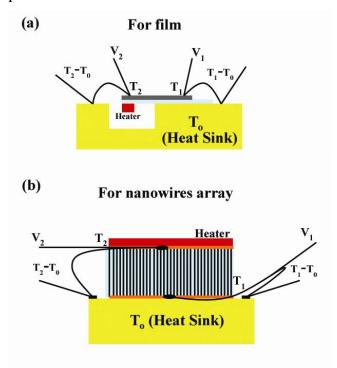


Figure B5. Schematic setup for Seebeck coefficient measurements on (a) film and (b) nanowires array in AAM.

Thermal diffusivity measurements were carried out using a laser flash equipment (LFA 457, NETZSCH) on flat disk samples [15]. Briefly described, the front side of specimen was heated by a short pulse of an Nd:YAG laser. The temperature on opposite side was measured by an infrared detector. The measured exothermic curve and laser pulse shape were evaluated by a fitting procedure [16,17]. Before each measurement, a piece of ultra-pure iron (99.999 %) was used as standard to assure the reliability of equipment. A differential scanning calorimeter (Q 100, TA instrument) was employed for heat capacity measurements.

Electrical conductivity σ can be derived from experimentally determined resistivity. The temperature dependence of resistance of nanowires array shows a much larger value than that expected, indicating that most of the nanowires have broken mid-sections inside the template. In order to obtain intrinsic electrical resistivity of nanowires, we set up a measurement on a single nanowire separated from the array at 300K, as shown in the inset of Figure B6a. E-beam lithography and thermal evaporation techniques were applied to fabricate four nickel electrodes onto a 120-nm diameter and 3.7-µm long nanowire. The perfectly linear current-voltage curve in Figure B6a yields a resistance of 1.42 k Ω . The corresponding resistivity of 4.3 $\mu\Omega$ -m is smaller than 19 $\mu\Omega$ -m of stoichiometric bulk Bi₂Te₃. This is somewhat expected for our Te-rich composition. According to Fleurial et al., the electrical resistivity of n-type bulk single-crystal Bi₂Te₃ decreases towards a low value of 2 $\mu\Omega$ -m with increasing Te content [18]. Meanwhile, theoretical studies also show that any vacancy, antistructural defects, and substitutional defects will produce higher polarization in bonding between the defect and neighbouring atoms, thereby contribute extra charges to conduction [19,20]. Figure B6b summarizes the temperature dependence of resistivity of each sample between 5 and 350 K. The film has its 300-K resistivity reduced considerably through annealing to 8.3 $\mu\Omega$ -m from 47.5 $\mu\Omega$ -m in the highly-disordered, as-prepared condition. It also displays complex temperature dependence, likely reflecting the interplay between semiconducting and semimetallic character. Indeed, even the normally semimetallic Bi₂Te₃ in the bulk form can become semiconducting by changing synthesis conditions or relative contents between Bi and Te [21,22]. Furthermore, the highly anisotropic structure could have different electronic and transport behaviors in different directions. different temperature dependence could likely contribute to the observed slope change at 200 and again at 50 K.

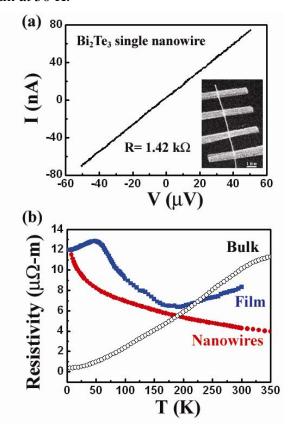


Figure B6. (a) I-V curve of single nanowire at 300 K. Inset: SEM image of four-probe setup. (b) Temperature dependence of electrical resistivity for bulk, annealed film and asprepared nanowires array.

The resistivity of nanowires array plotted in Figure B6b was normalized by the magnitude of single nanowire measured at 300 K. In contrast to the bulk, a semiconducting-type behavior prevails in nanowires array. Its resistivity, along the nanowire axis, follows an exponential temperature dependence. This observation, however, disagrees with theoretical calculations, which result in a positive temperature dependence of resistivity for polycrystalline Bi₂Te₃ nanowires [23]. The discrepancy is not surprising, considering the off-stoichiometry and high degree of disorder in the asprepared nanowires. In fact, such variations in resistivity are often characteristic of nanocrystalline materials, which possess different degrees of disorder and formation of grain boundary barriers in microstructure [24].

Seebeck coefficient data are essential to evaluate thermoelectric materials. Figure B7 shows the temperature dependence of Seebeck coefficients of Bi₂Te₃ bulk, annealed films and as-prepared nanowires from 120 to 350 K. The absolute value of S decreases almost linearly with decreasing temperature, demonstrating diffusive nature of current flow. Comparable results have been previously reported [25].

In the framework of two band model, $S = (\sigma_e S_e + \sigma_h S_h)/(\sigma_e + \sigma_h)$ [26]. For as-prepared films and nanowires along the nanowire axis, $S \approx -65 \mu V/K$ at 300 K, the negative sign points to electrons as majority carriers and the material as a n-type semiconductor. This relatively low value, being less than half of that of bulk, was probably a consequence of structural imperfections and excess Te in electrodeposited, nanocrystalline materials. Structural imperfections caused by anti-site defects [20] and additional Te are able to contribute more carrier concentration. There is often an optimal value of charge carrier density at which Seebeck coefficient is maximum [27]. Higher carrier concentration due to defects would result in a lower Seebeck coefficient. Our electrical resistivity data also suggest that the carrier concentration is likely beyond the optimal value and leads to a lower Seebeck coefficient. Annealing of as-prepared films has been shown to be able in reducing structural defects, as demonstrated by the significant reduction of electrical resistance. However, it raises Seebeck coefficient of films only slightly from -65 to -80 µV/K at 300 K, suggesting that the type of defects critical to S cannot be simply improved by annealing. Nevertheless, using resistivity and Seebeck coefficient data thus obtained from annealed film, a power factor (σS^2) of 860 μ W/m K² at 300 K is derived. This value sits among those of best electrodeposited films. In the as-prepared nanowires, power factor is estimated to be $\approx 1000 \,\mu\text{W/m} \,\text{K}^2$, about 15% superior to that of annealed film.

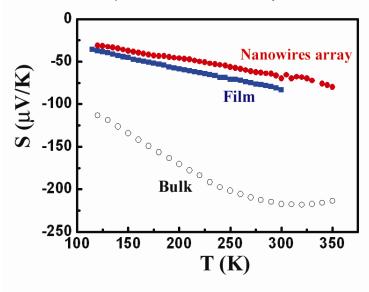


Figure 7. Temperature dependence of Seebeck coefficient of Bi₂Te₃: bulk, annealed film and asprepared nanowires array.

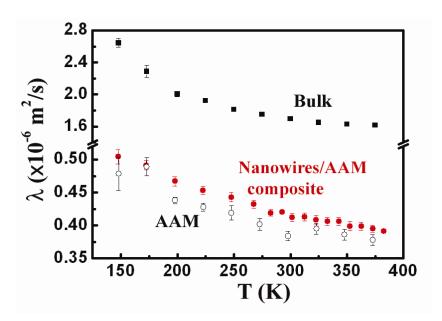


Figure B8. Temperature dependence of thermal diffusivity of unfilled AAM and Bi₂Te₃: bulk and nanowires/AAM composite

Instead of direct measurements, thermal conductivity can normally be determined from the relation:

$$\kappa = \lambda \rho C_p$$

where λ , ρ and C_p are thermal diffusivity, density and heat capacity, respectively. In turn, thermal diffusivity was measured using a laser flash apparatus (NETZSCH). The results are summarized in Figure B8. For bulk Bi₂Te₃, the various parameters ($\lambda = 1.69 \times 10^{-6}$ m²/s, $\rho = 7.70$ g/cm³ and $C_p = 0.16$ J/g K) yield $\kappa = 2.08$ W/m K at 300 K. Similarly, for unfilled AAM template along the channel axis ($\lambda = 0.38 \times 10^{-6}$ m²/s, $\rho = 3.35$ g/cm³ and $C_p = 0.77$ J/g K), $\kappa = 0.98$ W/m K at 300 K. When it is applied to nanowires/AAM composite along the nanowire axis, a λ of 0.41×10^{-6} m²/s at 300 K was obtained, which is slightly higher than that of unfilled AAM template, but only about one-fourth of bulk value. Unfortunately, from this result one cannot generate the thermal conductivity of the nanowires alone. To overcome this difficulty, we employ an alternative approach based on a modified effective medium theory model [16,28]. In this model, thermal diffusivity of the composite λ_{NW+AAM} along the nanowire axis has the following dependence on various parameters of Bi₂Te₃ nanowires and AAM, respectively:

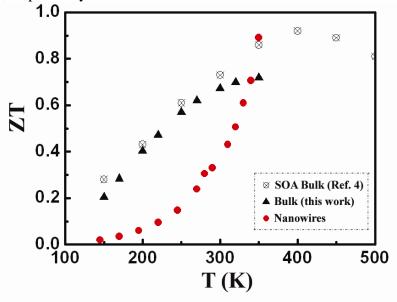


Figure B9. Temperature dependence of ZT for nanowires and bulks

$$\lambda_{\text{NW+AAM}} = \frac{(1-\Phi) \cdot \kappa_{AAM} + \Phi \cdot \kappa_{NW}}{(1-\Phi) \cdot (\rho C_{\rho})_{AAM} + \Phi \cdot (\rho C_{\rho})_{NW}}.$$

The percentage of nanowire in the composite, $\Phi \sim 25$ %, was estimated by the cross-sectional area ratio of the nanowire to the whole composite. The diameter of nanowires ranges between 110 and 135 nm, with an average of about 120 nm. If we assume that nanowires have the same properties as those of bulk, the calculated thermal diffusivity $\lambda_{\text{NW+AAM}}$ of 0.56×10^{-6} m²/s would obviously be overestimated as compared to the measured value of 0.41×10^{-6} m²/s at 300 K. To make the two numbers match, we arrive at a smaller $\kappa = 0.75$ W/m K for nanowires, which is about one-third of that of bulk. The considerable reductions in thermal conductivity for nanowires were also observed by other groups [29,30]. Then, for a single nanowire with the same $\kappa = 0.75$ W/m K and a power factor of $1000~\mu\text{W/m}$ K², we estimated the figure of merit ZT was about 0.45 at 300 K.

Finally, temperature dependence of figure of merit for nanowires between 150 and 375 K in Figure B9 was derived from measured S, σ and κ at each given temperature T. As compared to the state-of-the-art (SOA) bulk Bi₂Te₃, its ZT monotonically increases with temperature and reaches to a maximum value of 1 at 400 K follows by a decreasing value at higher temperature, it can be seen that the ZT of nanowires increases dramatically with temperature, and reaches 0.9 at 350 K. It looks like the ZT can increase to a much higher value if the measurement can be performed at higher temperature.

B4. Summary of the results

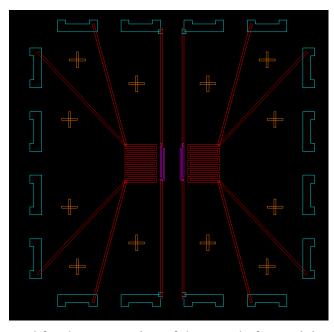


Figure C1. The template used for the preparation of the sample for studying thermal rectification effect. The pattern includes contacts and leads for electrical property measurements; and heater for thermal properties characterizations.

Bismuth telluride film and nanowires array (embedded in alumina template) were fabricated by potentiostatically electrodeposition. Both materials are slightly Te-rich, n-type Bi_2Te_3 , exhibiting preferred orientation in rhombohedral strcture. Their Seebeck coefficient decreases linearly from -70 μ V/K at 300 K with decreasing temperature, showing a diffusive nature of current flow. Electrical resistivity of a single nanowire separated from the array was measured at 300 K, which was then used as reference in determining resistivity of nanowires at other temperatures. Meanwhile, from temperature-dependent thermal diffusivity and heat capacity data, a thermal conductivity value of 0.75 W/m K was obtained at 300 K. These parameters subsequently provide an estimated thermoelectric figure of merit ZT = 0.45 at 300 K and likely to reach beyond 0.9 above 350 K, for

Bi₂Te₃ nanowires. It can be thus concluded that significantly enhanced ZT values can likely be achieved in Bi₂Te₃ nanowires at the relatively high temperature region.

Research approach C—Directional dependent thermal conductivity and thermal rectifier

In order to demonstrate such an effect with a controllable device, we have designed and developed a unique structure to prepare the samples, including the specific template for properties characterizations. Figure C1 shows the lithographic pattern we used in this study. Figure C2 is the fabricated pattern. A tapered feature is used.

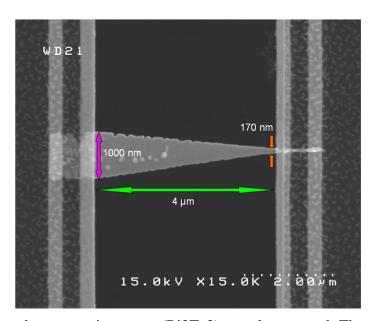
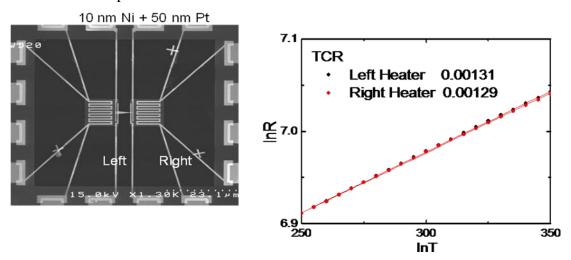


Figure C2. The fabricated asymmetric pattern (Bi2Te3) sample prepared. The width of the narrowest region is ~ 170 nm, the widest region is ~ 1000 nm and the length is $\sim 4 \mu m$. It is clearly observed from the SEM image that the edges of the tapered pattern exhibiting zig-zag features.

In order to characterize the properties of the fabricated pattern we need to setup a program to sweep the current with different directions, and then to determine whether there exists any difference. Figure C3a shows the sample pattern with the two heaters made of 10 nm Ni and 50 nm Pt. Figure C3b is the calibrated resistance vs. temperature with the heaters used. Figure C4 displays the schematic of the heat flow needs to be considered and the details how to analyse the thermal characteristics in this special device.



Fiure C3. The samples used in this study and the resistance-temperature relation of the heater used.

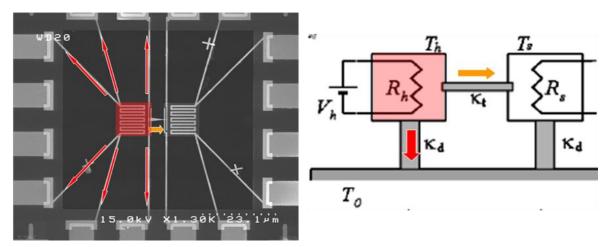


Figure C4. The schematic diagram showing the heat flow in this special device and the parameters to be determined.

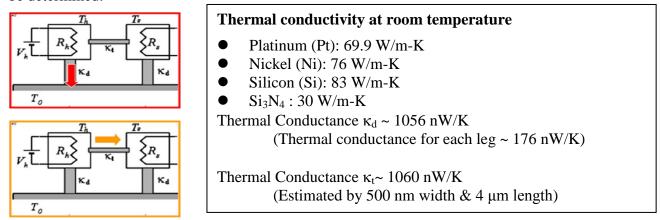


Figure C5. Schematic diagrams the material values used to estimate thermal conductance.

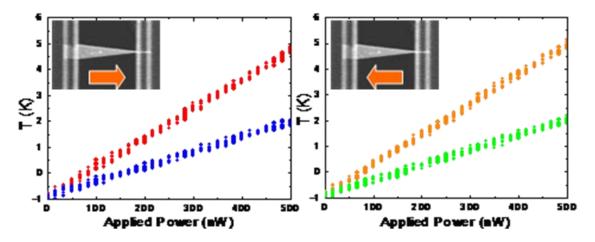


Figure C6. The measured temperature vs. applied power of the tapered device. The left panel is the result of heat flows from left region to the right; whereas the right panel is the reverse case.

To evaluate the thermoelctric effect we first need to know the thermal conductance of the device. Consider the system as schetch in figure C4, it is clear that one need first to find out the thermal conductance κ_t and κ_d , which represent the thermal conductance of the tapered structure and the conductance to the heat bath, respectively. Based on the known material parameters, as listed in figure C5, we estimate $\kappa_t \sim 1060$ nW/K and $\kappa_d \sim 1056$ nW/K. With these estimated thermal

conductance we are then able to determine the change in temperature by the applied power. The data shown in figure C6 are the results. The two different curves in each panel are the change along κ_t (red—left to right heat flow; or orange—right to left) and κ_d (blue—left to right heat flow or green—right to left) respectively.

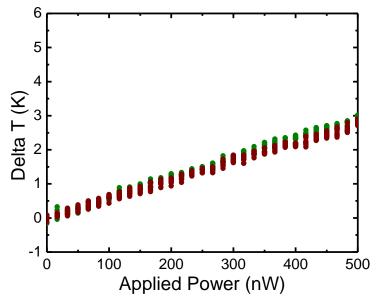


Figure C7. ΔT versus applied power of the tapered device investigated. The data show the temperature changes in the two directions (left-to-right and right-to-left) are the same within the experimental error.

The final value of the temperature change ΔT of the device as a function of the applied power can be sumarrized in figure C7. We plot in this figure both the changes from left to right and vice versa. The results suggest that the difference in the heat flow in forward and reverse direction is very small. However, we cannot make a conclusive statement that there exists no rectification effect, as the narrowest section in our device is still too big compared with that observed in carbon and boron nitride nanotubes [31].

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Publications:

In addition to the works on thermoelectric effect as originally proposed, we have also benefit from this research grant an outstanding research on superconductivity, especially in the area of the newly discovered Fe-based superconductors. The followings are the list of our publications during the period under this grant support.

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